

**PATENT**  
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**UNITED STATES PATENT APPLICATION**

for

**CARBON NANOTUBE BASED RESONANT-CIRCUIT SENSOR**

of

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and

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**CARBON NANOTUBE BASED RESONANT-CIRCUIT SENSOR**

**Federally Sponsored Research and Development**

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**Background of the Invention**

10           [0002] Sensors are used in countless applications throughout a myriad of industries based upon their ability to respond in a measurable fashion to a stimulus. Sensors are used to measure or respond to everything from the presence of hazardous liquids or gases in a manufacturing process to the current weather conditions anywhere on the planet to stolen goods passing out the door of a retail establishment. With the advent of increasingly automated manufacturing  
15           processes as well as the possibility of hazardous environmental conditions caused by natural or induced disaster, gas sensors, such as may be utilized to discern the presence of gas or vapor-born pathogens, have attracted increasing attention of late.

20           [0003] In general, gas sensors are those sensors which convert a chemical signal to an electrical signal, with a change in the chemical input leading to a measurable change in the electrical output. Traditionally, gas sensors have been formed using semi-conducting oxides as the sensing material. These materials require expensive microfabrication techniques to form the sensors, however, and often only have the high sensitivities required to discern materials in low  
25           concentrations (such as parts per million) at high temperature (e.g., 200–600°C).

30           [0004] Recently, carbon nanotubes have been examined as possible gas sensing materials due to their electrical and mechanical properties as well as their high specific surface area. For example, chemical sensors based on individual nanotubes have been reported by Kong, et al. ("Nanotube Molecular Wires as Chemical Sensors," *Science*, 287, 622 (2000)). These sensors utilize the measurable change in the electrical resistance of a nanotube upon exposure to gases like NO<sub>2</sub> and NH<sub>3</sub> to sense the presence of the gas.

[0005] Inductor-capacitor (LC) resonant sensors are a type of solid-state sensor based upon the permittivity of a material. LC resonant sensors have been widely utilized as remote anti-theft sensors. In these remote sensing systems, an RF transmitter/receiver sends a microwave signal at a targeted frequency through an interrogation zone. The presence in the interrogation zone of an activated tag including a dielectric material having the targeted resonant frequency can be detected by the receiver, which then sets off an alarm.

[0006] More recently, research has been carried out to expand the use of LC resonant sensors beyond these simple yes- or no-type applications. Specifically, LC resonant sensors have been combined with carbon nanotube materials for utilization as gas sensors. For example, Ong, et al. ("A Wireless, Passive Carbon Nanotube-Based Gas Sensor," IEEE Sensors Journal, Vol. 2, No. 2, April, 2002) has described a gas sensor formed of a responsive multi-wall carbon nanotube/silicon dioxide composite layer deposited on a planar LC resonant circuit. As the permittivity and/or conductivity of the MWNT/SiO<sub>2</sub> composite changes with adsorption of CO<sub>2</sub>, O<sub>2</sub>, or NH<sub>3</sub>, so does the resonant frequency of the sensor, which can be remotely monitored through a loop antenna. The sensors showed reversible response to O<sub>2</sub> and CO<sub>2</sub>, and an irreversible response to NH<sub>3</sub>.

[0007] Chopra, et al. ("Carbon-nanotube-based Resonant-circuit Sensor for Ammonia," Applied Physics Letters, Volume 8, Number 24, 2002, which is incorporated herein in its entirety by reference thereto) have described an ammonia sensor formed of a simple micro-strip circular disk resonator coated with carbon nanotubes (either single-walled or multi-walled nanotubes) on the surface. The sensors show a shift in resonant frequency upon adsorption of ammonia of about 4.375 MHz for a single-walled nanotube (SWNT) sensor and a shift of about 3.25 MHz for a multi-walled nanotube (MWNT) sensor, and can detect the presence of ammonia down to a concentration of about 100 ppm.

[0008] Despite these advances in addressing the needs for improved gas sensors, there remains room for variation and improvement within the art.

### **Summary of the Invention**

[0009] The present invention is directed to resonant circuit gas sensors and methods of forming and utilizing the disclosed gas sensors. For instance, in one

embodiment, the present invention is directed to a gas sensor including a resonator comprising a dielectric material and a layer including an amount of adsorptive nanostructures applied to a surface of the resonator, wherein this layer is in electrical communication with the dielectric material. That is, the

5 electromagnetic field coupling characteristics of the resonator have been relaxed from that of a perfectly matched resonator such that electromagnetic field lines can couple to both the layer of the sensor that includes the adsorptive nanostructures as well as through the dielectric substrate. As such, the effective resonant frequency of the gas sensor can depend not only on the dielectric constant of the dielectric material, but also on the dielectric constant of the layer comprising the  
10 adsorptive nanostructures. In addition, the dielectric constant of this layer can vary in proportion to the dielectric of a material adsorbed onto or in electrical communication with the nanostructures. The adsorptive nanostructures can be, for example, carbon nanotubes, activated carbon fibers, or adsorptive nanowires.

15 [0010] In one embodiment of the invention, the adsorptive nanostructures can be degassed carbon nanotubes. In this embodiment, the carbon nanotubes can be degassed by a process including holding the resonator at low pressure and high temperature for a period of time following formation of the resonator. For example, the resonator can be held at a pressure between about  $1 \times 10^{-5}$  and  $2 \times 10^{-4}$   
20 torr for a period of at least about 12 hours while simultaneously holding the resonator at a temperature of between about 100°C and about 125°C.

[0011] The disclosed sensors can also include an analyzer for obtaining the resonant frequency of the resonator. Moreover, the analyzer can be either hard wired to the resonator or in remote communication with the resonator, such as via  
25 a radio frequency signal.

[0012] In one embodiment, the adsorptive nanostructures can be carbon nanotubes. In this embodiment, the carbon nanotubes can be either single-walled nanotubes (SWNTs) or multi-walled nanotubes (MWNTs), as desired. For example, in one embodiment, a layer of carbon nanotubes can be applied to a  
30 surface of the resonator and can include SWNT bundles. Individual SWNT bundles can have, in one embodiment, diameters between about 5 and about 200 nanometers. In another embodiment, the layer of nanotubes applied to a surface

of the resonator can include MWNTs. In general, MWNTs can have diameters between about 5 and about 100 nanometers.

[0013] The depth of the layer comprising adsorptive nanostructures can vary. For example, in one embodiment, the layer of the sensor including the  
5 adsorptive nanostructures can be about 2  $\mu\text{m}$  in depth.

[0014] In one particular embodiment, the resonator of the disclosed sensor can be a micro-strip circuit board resonator. In this embodiment, the resonator can include a ground plane, a dielectric substrate applied to the ground plane, a  
conducting micro-strip applied to the dielectric substrate, and a layer comprising  
10 adsorptive nanostructures, such as degassed carbon nanotubes, applied to a surface of the conducting micro-strip. In one particular embodiment, the conducting micro-strip can be a circular disk.

[0015] When utilizing the disclosed sensors, in one embodiment, the resonator of the sensor can be contacted with a gaseous stream. Material in the  
15 stream can adsorb to the adsorptive nanostructures on the resonator, which can cause a change in the effective dielectric constant of the resonator that can be reflected by a change in the resonant frequency of the resonator. Beneficially, the sensors can be utilized at room temperature to detect very low concentrations of materials.

[0016] In one embodiment, the disclosed sensor can be utilized to detect the  
20 presence of a vaporous organic material in a gaseous stream. In one particular embodiment, the organic material can be adsorbed onto a layer comprising adsorptive nanostructures including as-prepared carbon nanotubes, that is, in this embodiment, the nanostructures can be carbon nanotubes that have not been  
25 degassed prior to use.

[0017] In some embodiments of the present invention, the adsorptive nanostructures can be degassed prior to use. In this embodiment, the sensitivity and selectivity of the sensors can be such that the sensors can detect the presence of polar gases, non-polar gases, organic vapors, or mixtures of such at  
30 very low concentrations, for example down to about 100 ppb. For instance, in certain embodiments, the sensors can detect materials at concentrations between about 100 ppb and about 1500 ppm.

[0018] The disclosed sensors can have very fast response times, on the order of about 10 minutes. In addition, the sensors can, in some embodiments, have recovery times also on the order of about 10 minutes.

### **Brief Description of the Drawings**

5 [0019] A full and enabling disclosure of the present invention, including the best mode thereof, to one of ordinary skill in the art, is set forth more particularly in the remainder of the specification, including reference to the accompanying figures, in which:

[0020] Figure 1 is a schematic diagram of an electric arc discharge  
10 apparatus such as may be utilized in forming SWNTs of the present invention;

[0021] Figure 2 is a schematic diagram of a laser-vaporization apparatus such as may be utilized in forming SWNTs of the present invention;

[0022] Figure 3 is a schematic diagram of a CVD apparatus such as may be utilized in forming MWNTs of the present invention;

15 [0023] Figure 4 illustrates one embodiment of a portion of a resonant-circuit sensor according to the present invention;

[0024] Figure 5 is a block diagram of a network analyzer suitable for use with the resonant-circuit sensors of the present invention;

[0025] Figure 6 is a schematic diagram of the testing apparatus used in the  
20 Examples;

[0026] Figure 7 is a schematic diagram of one embodiment of the disclosed sensors in a remote sensing application; and

[0027] Figures 8-14 graphically illustrate the responses of resonant-circuit sensors of the present invention to various gases and organic solvent vapors.

25 [0028] Repeat use of reference characters in the present specification and drawings is intended to represent the same or analogous features or elements of the present invention.

### **Detailed Description of the Invention**

[0029] Reference will now be made in detail to various embodiments of the  
30 invention, one or more examples of which are set forth below. Each embodiment is provided by way of explanation of the invention, not limitation of the invention. In fact, it will be apparent to those skilled in the art that various modifications and variations may be made in the present invention without departing from the scope

or spirit of the invention. For instance, features illustrated or described as part of one embodiment, may be used in another embodiment to yield a still further embodiment. Thus, it is intended that the present invention cover such modifications and variations as come within the scope of the appended claims and their equivalents.

[0030] The present invention is generally directed to gas/vapor sensors that are able to detect the presence of a large number of different substances at very low concentration levels. The invention is also directed to methods for forming the disclosed sensors as well as methods for utilizing the disclosed sensors. More specifically, the presently disclosed sensors can detect the presence of both polar and non-polar gases. In addition, the disclosed sensors can detect the presence of organic vapors. Beneficially, the disclosed sensors display both high sensitivity, able in some embodiments to detect the presence of materials at concentrations on the order of 100 parts per billion (ppb), as well as high selectivity, showing measurably different responses to a wide variety of different materials. Moreover, the disclosed sensors can operate at room temperature and can have fast response and recovery times, on the order of about 10 minutes. In addition, in certain embodiments, the disclosed sensors can completely recover their original electrical state following the recovery procedure, with little or no residual effects on the sensor due to previous use. In one embodiment, the disclosed sensors can operate in remote sensing systems, in which the analyzer and the material sensing portion of the sensor are in remote contact, such as through an RF transmitter/receiver, for example. Alternatively, in other embodiments, the analyzer and sensing portions of the sensors can be in hard wired communication.

[0031] The basic principle of gas/vapor sensors is the conversion of a chemical signal (the presence of the gas or vapor) to an electrical signal. The presently disclosed sensors are microwave resonant-circuit-type sensors in which exposure to a gas or vapor can cause a change in the dielectric behavior of the sensor, which can be demonstrated by variation in the resonant frequency of the sensor.

[0032] In general, a microwave resonator can be thought of as a high frequency substitute of a lower frequency lumped circuit LC network as is generally known in the art. A conventional microwave resonator has a bounded

electromagnetic field in a dielectric material that is encased by metallic walls. The electric and magnetic energies are stored in the electric and magnetic fields inside the dielectric and the corresponding lumped circuit equivalent inductances and capacitances can be calculated from the stored fields. Thus, the resonant frequency of any particular resonator will depend not only on the geometry of the resonator, but also on the effective dielectric constant of the dielectric materials. For example, when considering a circular disk resonator, the resonant frequency of the resonator can be approximated (by neglecting the edge fields of the disk) by the expression:

10 
$$f = \frac{1.841 c}{2\pi a \sqrt{\epsilon_r}}$$

where  $c$  is the velocity of light,  $\epsilon_r$  is the relative dielectric constant of the substrate,  $a$  is the radius of the circular disk and  $f$  is the resonant frequency of the resonator.

[0033] In the present invention, the resonator portion of the sensor can include several different materials, all of which can contribute to the electrical characteristics, and specifically to the effective dielectric constant, and hence the resonant frequency, of the sensor. In particular, the disclosed sensors include a layer including adsorptive nanostructures at a surface of the resonator. Suitable adsorptive nanostructures can be any structure formed on a nano-scale that can adsorb materials from a gaseous state, i.e., gases or vapors. Adsorptive nanostructures can include, for example, carbon nanotubes (either single-walled nanotubes (SWNTs) or multi-walled nanotubes (MWNTs)), activated carbon fibers, adsorptive nanowires, and the like. Moreover, when considering carbon nanotubes, both functionalized and non-functionalized carbon nanotubes are encompassed by the invention. In addition, nanotubes with either open ends or closed end-caps can be utilized with the disclosed invention.

[0034] The design of the resonators of the disclosed sensors are such that the electromagnetic field coupling characteristics of the resonator have been relaxed from that of a perfectly matched resonator. In this manner, electromagnetic field lines can pass through the area of the sensor in which the adsorptive nanostructure-containing layer is located as well as through the dielectric substrate itself. Thus, the dielectric material and the adsorptive nanostructure-containing layer are in electrical communication with one another,



and the electrical characteristics of both the dielectric and the adsorptive nanostructure-containing layer can affect the electrical characteristics of the sensor, and specifically the resonant frequency of the sensor.

5 [0035] In the present invention, it is the ability of nanostructures to adsorb materials as well as the high specific surface area of the nanostructures which are of particular relevance. For instance, due to the high specific surface area of nanostructures, little nanostructure material is required on the sensors to provide ample sites for gas or vapor interaction, and the sensors of the present invention can, if desired, be quite small.

10 [0036] Carbon nanotubes in general can exhibit exceptional physical strength, elasticity, hydrogen storage capability, adsorption capability, high specific surface area, and resistance to most chemicals. As such, in one particular embodiment of the present invention, the adsorptive nanostructures of the disclosed invention can be carbon nanotubes. The following discussion is  
15 therefore directed to this particular embodiment. It should be made clear however, that though the following discussion is directed to one particular exemplary embodiment, any adsorptive nanostructures can equally be utilized in the disclosed invention, and in much of the following discussion, the term 'adsorptive nanostructure' is directly interchangeable with the term 'carbon nanotube'.

20 [0037] The ability of carbon nanotubes to quickly adsorb materials is of benefit to the disclosed sensors. More particularly, when considering the effective dielectric constant and associated resonant frequency of the resonators of the disclosed sensors, not only will the presence of the nanotubes on the resonator affect the resonant frequency, but the added presence of materials adsorbed onto  
25 the nanotubes will also affect the resonant frequency. In addition, as dielectric constant is an inherent characteristic of a material, the effective dielectric constant of the resonator, and hence the resonant frequency of the resonator, will vary depending on exactly what materials have been adsorbed onto the carbon nanotubes. Thus, a shift in resonant frequency of the resonator can be observed  
30 upon adsorption of a material to the nanotubes. Moreover, this shift will vary depending on what material is adsorbed. Beneficially, the ability of the carbon nanotube-containing layer to adsorb a material does not require the use of functionalized carbon nanotubes for enhanced selectivity to any particular

chemical species, and as such, the formation processes for the disclosed sensors can be relatively simple and inexpensive.

[0038] In certain embodiments of the present invention, the disclosed sensors can be made more sensitive by degassing the adsorptive layer of the resonator. In this particular embodiment, it has been found that the sensitivity of the sensors can be improved, providing sensors that can respond to the presence of gases and vapors in concentration levels as little as about 100 ppb. In addition, when the nanotube-containing material is degassed prior to use the sensors can have measurable response to an increased number of materials. For example, when utilizing degassed nanotubes in the sensors, the sensors can indicate a measurable variation in resonant frequency of the resonator upon exposure to polar as well as non-polar gases.

[0039] Referring to Figures 4A and 4B, one embodiment a resonator generally 10 that can be utilized in the disclosed sensors is illustrated. In this particular embodiment, the resonator 10 is a microwave circuit board resonator comprising a ground plane 18, a dielectric substrate 12 and a conducting micro-strip 16. In one embodiment, the ground plane 18 can be a metallic plate, such as a copper plate. A dielectric substrate 12 can be applied to the surface of the ground plane 18, as shown. The dielectric substrate 12 can be any suitable dielectric material as is known in the art. In one embodiment, the pre-shaped conducting micro-strip 16 can be applied to the surface of the dielectric substrate 12. In another embodiment, the conducting micro-strip 16 can be applied and then etched to the desired final geometry following application. In yet another embodiment, the resonator 10 can be formed from a Duroid board, i.e. a dielectric sandwiched between metal (usually copper) plates, in which the desired pattern for the micro-strip 16 has been formed and etched in the metal on one side of the board.

[0040] It should be understood that the specific circuit board materials, geometries, and production methods described for the microwave circuit board of this particular embodiment of the disclosed sensors are not critical to the presently disclosed invention, and other microwave circuit board materials and geometries as are generally known in the art may alternatively be utilized for any of the ground plane 18, the substrate 12, and/or the conducting micro-strip 16.

[0041] As can be seen, in the embodiment illustrated in Figures 4A and 4B, resonator 10 is a circular disk resonator, which is a type of micro-strip resonator as is generally known in the art. There are several different types of resonators and resonator geometries known in the art, however, and, in particular, as the effective dielectric constant will vary with the geometry of the resonator, among other factors, other types of resonators are also encompassed by the disclosed invention. For example, in other embodiments of the invention, different geometries of micro-strip resonators are contemplated, as previously discussed. In addition, however, other types of resonators could be utilized in the disclosed sensors including, for example, rectangular cavity resonators or dielectric resonators. In general, any type of resonator can be utilized in the sensor as long as the resonator defines at least one surface to provide suitable interaction between the sensor and the gaseous materials to be detected. In addition, the dielectric of the resonator must be in electrical communication with adsorptive nanostructures located on this surface.

[0042] According to the present invention, the resonator includes a layer including an amount of adsorptive nanostructures, such as carbon nanotubes, on a surface. For example, in the embodiment illustrated in Figures 4A and 4B, a layer 14 including carbon nanotubes can be physically coated or directly deposited on a surface of the conducting micro-strip 16. As can be seen, the layer 14 containing carbon nanotubes need not be in physical contact with the dielectric material 12 of the resonator, but the coupling characteristics of the resonator can be relaxed from that of a perfectly matched resonator such that the layer 14 containing carbon nanotubes can be in electrical communication with the dielectric material 12. As such, the electrical characteristics of the nanotube layer, e.g., the dielectric constant, the conductivity, etc. can affect the electrical characteristics of the resonator 10. Similarly, material adsorbed to the carbon nanotubes can also affect the electrical characteristics of the resonator. Thus, upon adsorption of a material to the carbon nanotubes, a measurable change in the electrical characteristics of the resonator 10 can be observed.

[0043] Carbon nanotubes that can be used in the disclosed sensors can be either SWNTs or MWNTs, as desired. In addition, the nanotubes can be formed according to any formation process as is known in the art. For example, in one

embodiment, the sensor can include a layer of SWNTs formed according to an electric arc method, one embodiment of which is illustrated in Figure 1. According to this method, an electric arc can be generated in an inert atmosphere, for example, helium at about 600 torr, between a cathode and an anode. In general, the cathode can be a graphite rod and the anode can be a catalyst-impregnated graphite rod, such as those available from Carbolex, Inc. of Lexington, Kentucky. For example, the catalyst can be a mixture of 1 at. % Yttrium and 4.2 at. % Nickel. An arc discharge can be created by a current of about 100 A, creating a voltage of about 25 volts between the two electrodes. Typical synthesis times are of the order of two hours and result in a web-like material containing the SWNTs between the cathodes and the reactor walls. SWNTs produced according to this method can have a diameter of about 1.4 nm and are generally several microns in length. The electric arc method for formation of the SWNTs of the invention may be preferred in some embodiments due to high yields, up to about 70%.

[0044] In another embodiment, SWNTs can be utilized in the invention formed according to the laser-vaporization method, an example of which is illustrated in Figure 2. According to this method, a laser beam, controlled by a motor driven reflector, can be focused on a rotating target about one inch in diameter. The target is a metal/graphite composite and can be placed at the center of a furnace. The furnace can be maintained at a high temperature, about 1100°C, for example. The laser beam scans laterally across the target and the target is ablated uniformly as it rotates, as shown by the arrow in Figure 2. The process is carried out in an inert atmosphere, and the material produced by the laser vaporization of the target is swept by inert flow out of the furnace and collected in a lower temperature zone, such as on a water-cooled condenser. In general, SWNTs produced according to the laser-vaporization method can be larger than those produced according to the electric-arc method.

[0045] In certain embodiments of the present invention, the carbon nanotube-containing layer of the sensor can include multi-walled carbon nanotubes (MWNTs). In one embodiment, MWNTs can be formed according to a chemical vapor deposition method, such as that illustrated in Figure 3. In the embodiment shown in Figure 3, highly aligned and high purity MWNTs can be produced by the thermal decomposition of a xylene-ferrocene mixture. The xylene

serves as the hydrocarbon source and ferrocene provides the iron catalyst nanoparticles that can seed the nanotubes that are grown. According to one process, ferrocene (approximately 6.5 at. %) can be dissolved in xylene and fed into a quartz tube at a flow rate of about 1 ml/hr. The mixture can vaporize upon reaching the end of the pre-heater (maintained at about 200°C), and the vapors can then be carried into the furnace in an Ar/H<sub>2</sub> flow. The furnace is maintained at a temperature (e.g., about 750°C) that enables the xylene/ferrocene mixture to decompose and form the MWNTs. The nanotubes are harvested from the walls of the furnace and, in one embodiment, can have a diameter of about 25 nm.

[0046] It should be understood that the specific method of forming the nanotubes of the disclosed sensors is not critical to the invention, and the described methods are merely exemplary, and not meant to be limiting in any way to the invention. In general, SWNT bundles utilized in the disclosed sensors can be between about 5 and about 200 nanometers in diameter, and MWNTs utilized in the disclosed sensors can be between about 5 and about 100 nanometers in diameter.

[0047] Referring again to Figure 4, in this embodiment, following formation, the nanotube-containing material can be coated on the surface of the conducting micro-strip 16 to form a layer 14 on a surface of the resonator 10. For example, in one embodiment, the nanotubes can be coated on top of the conducting micro-strip 16 using conductive epoxy such as CW2400 available from Circuit Works of Lake Bluff, Illinois. In another embodiment, the nanotube material can be directly deposited on the resonator. For instance, in one embodiment, the carbon nanotube-containing material can be directly deposited on the conducting micro-strip during the nanotube formation process such that the carbon nanotubes can be directly grown on the surface of the conducting micro-strip of the resonator.

[0048] The depth and purity of the nanotube-containing layer 14 on the sensors is not critical to the invention. In particular, all that is necessary is that suitable amounts of nanotube material be coated on the sensor so as to provide an amount of surface area for interaction with the materials to be detected. For example, in one embodiment, the disclosed sensor can include an adsorptive nanostructure-containing layer that is about 2μm thick.

[0049] The sensors of the present invention can be either hard-wired sensors or remote sensors. For example, in the embodiment illustrated in Figures 4A and 4B, the sensor includes a hard wire connector 20 to an analyzer and an input feed line 22 that can couple the energy fed to the device through the connector into the resonator 10 through the dielectric substrate 12. For example, the connector 20 can be a hard wire connection such as a coaxial cable connected to an analyzer.

[0050] In another embodiment, illustrated in Figure 7, the presently disclosed sensor can operate as a remote sensor. In this embodiment, the sensor can include an analyzer 30 that includes a transmitter/receiver antenna 32. During use, a microwave signal 34 transmitted from analyzer 30 can interact with the resonator via input receiver 23 coupled to input feed line 22 and monitor the resonant frequency of the resonator. Antenna 32 of the analyzer 30 can detect the return signal 35 at the resonant frequency of the resonator 10, and any resonant frequency shift upon adsorption of a material can be analyzed in a like manner to the hard wired system. Beneficially, in this embodiment, the sensor can be utilized in applications in which there may be intervening materials between the analyzer and the sensor. For example, in one embodiment, the enclosed chamber 40 can be an enclosed, air-tight packaging system, such as commonly used for food or medical products. In this embodiment, the sensor can detect the presence of gases or vapors within the packaging that could signal damaged goods.

[0051] Figure 5 illustrates a block diagram for one embodiment of a network analyzer generally 30 that can be used in conjunction with the resonator of the disclosed invention in either a hard wired or remote access configuration to obtain and manifest the resonant frequency of the resonator. In general, a suitable network analyzer 30 can include a sweeping signal source, a test set to separate forward and reverse signals, and a multi-channel, phase-coherent, highly sensitive receiver. As analysis methods and instrumentation necessary for the analyzer of the disclosed sensors is generally known in the art, a detailed description of such is not provided herein.

[0052] Utilizing suitable analysis methods and instrumentation, the resonant frequency shift of the resonator upon adsorption of a material to the carbon nanotubes contained in a layer of the resonator can be detected and analyzed and

the adsorbed material can be identified based upon the measured shift in resonant frequency. In one embodiment of the present invention, the sensor can also detect and analyze the change in the Q-factor (loss in conductivity) of the resonator upon adsorption of a material in addition to the change in the resonant frequency. While  
5 not required by the disclosed sensors, the analysis of the Q-factor can serve as an additional variable to monitor the presence of adsorbed materials on the nanotube-containing layer of the resonator.

[0053] The resonant sensors of the disclosed invention can be used to detect a variety of different materials at very low concentrations. In particular,  
10 because the disclosed resonant sensors operate in the microwave regime, a very small change in the effective dielectric constant of the sensors, which can be generated in some embodiments by the presence of materials in as low as parts per billion concentrations, can lead to a discernable shift in the resonant frequency of the resonator. In addition, at low concentrations, and depending upon the  
15 nature of the detected material, the shift in resonant frequency can be proportional to the concentration of the detected material. As such, in certain embodiments, the disclosed sensors can also be utilized to analyze the concentration of a detected material.

[0054] As previously discussed, the effective dielectric constant of the  
20 resonator can vary depending upon the different materials in and on the resonator. Thus, when a material is adsorbed onto the carbon nanotubes, the effective dielectric constant of the resonator can shift and a corresponding shift in resonant frequency can be detected and analyzed. When the resonator of the sensor is formed and utilized as-prepared, that is, in air, this will naturally lead to a certain  
25 amount of material adsorbing to the nanotubes from the air. For example, oxygen, nitrogen and argon can be expected to adsorb to the nanotube layer of the sensor when the resonator is in air. This can shift the effective dielectric constant of the resonator. As such, the resonator of the present invention, when utilized as-prepared, under atmospheric conditions, may be 'blind' to certain materials, and  
30 particularly to certain non-polar and inert materials. In addition, the presence of materials previously adsorbed onto the nanotube layer can decrease the sensitivity of the resonator during sensing operations. As such, in one embodiment, the sensitivity and selectivity of the disclosed resonator sensors can be increased by

degassing the nanotubes prior to use and utilizing the sensors with the resonator held in a degassed state.

[0055] Generally, the nanotubes held in the nanotube-containing layer can be degassed following formation of the resonator by holding the resonator at an elevated temperature for a period of time at low pressure. For example, in one embodiment, a resonator can be held at a pressure between about  $1 \times 10^{-5}$  torr and about  $2 \times 10^{-4}$  torr and simultaneously at a temperature of between about  $100^{\circ}\text{C}$  and about  $125^{\circ}\text{C}$  for at least about 12 hours to degas the nanotubes of the resonator. Following the degassing procedure, the resonator can be held at low pressure during cooling and utilization to prevent re-adsorption of atmospheric materials on the resonator.

[0056] Through the process of degassing the nanotubes, it has been found that not only can the sensors be utilized to detect materials to which they were previously 'blind' including non-polar and inert gases, for example, but in addition, the sensitivity of the sensors can be greatly increased. For example, when considering the detection of the polar gas ammonia, when utilizing a resonant sensor of the disclosed invention including a layer of SWNT as-prepared (that is, formed in air and not degassed following formation) the sensor can detect the presence of ammonia down to a concentration of about 100 ppm. In contrast, following a degassing procedure such as that outlined above, the disclosed sensors can detect the presence of ammonia down to a concentration of about 100 ppb.

[0057] In addition to the high sensitivity and selectivity of the disclosed sensors, the sensors of the present invention can also detect the presence of materials over a wide concentration span. For example, the disclosed sensors can detect the presence of polar gases, non-polar gases, and organic vapors in concentrations ranging from about 100 ppb to at least about 1500 ppm, or even higher in some embodiments, without saturating the sensor. This is believed to be primarily due to the high specific surface area of the nanotubes held in a layer of the resonator.

[0058] The sensors of the disclosed invention can have relatively fast response times to the presence of a material in a gaseous stream (i.e., a gas or vaporized material). For example, in certain embodiments, upon contact with a



gaseous material, the disclosed sensors can exhibit a measurable change in resonant frequency within about ten minutes. In addition, the disclosed sensors can also exhibit a fast recovery time. For example, in those embodiments wherein as-prepared nanotubes are applied to the resonator and utilized for the detection of materials, such as polar gases like ammonia or organic vapors, the initial resonant frequency of the resonator can often be recovered within about ten minutes merely by evacuating the testing chamber of the gaseous material. When utilizing degassed nanotubes, the recovery period can be slightly longer, however, as the recovery of the initial resonant frequency of the resonator can often require a degassing process, in order to remove any molecules chemisorbed to the surface of the nanotube layer. Thus, in certain embodiments, complete recovery of the sensor can take somewhat longer, on the order of a few hours.

[0059] The disclosed invention can be better understood with references to the following examples.

#### Example 1

[0060] A circular disk resonator such as that illustrated in Figures 4A and 4B was formed. The circular disk resonator included a Duroid Board (Rogers RO4003) that was milled to fabricate a circular disk resonator resonating at 4GHz. The surface of the micro-strip disk was coated to a depth of about 2  $\mu\text{m}$  with a layer including single-walled nanotubes formed according to an arc-discharge process as illustrated in Figure 1. The SWNT-containing layer was coated on the micro-strip disk using a conductive epoxy (CW2400, Circuit Works) as the adhesive. This resonator sample was then placed in a testing chamber as illustrated in Figure 6. The response of the sensor was monitored under different gas environments.

[0061] Figure 9 shows the response of the resonator under helium. The solid line shows the response of the resonator in air. In air, the sensor displayed a resonant frequency of 3.887 GHz. The chamber was then pulled down to high vacuum ( $\sim 10^{-5}$  torr) and heated to  $\sim 125^{\circ}\text{C}$  for about 12 hours to degas the SWNTs. The testing chamber was then allowed to cool to room temperature. The second plot (squares) on Figure 9 shows the sensor response after degassing. Following degassing, the resonator had a resonant frequency of 3.8893 GHz, having shifted from the previous value by 2.3 MHz ( $\Delta f_0$ ). The resonator was then exposed to

known amounts (~1500 ppm) of helium gas and the response of the sensor was measured after 10 minutes (circles). The resonant frequency shifted to 3.8885 GHz with a decrease of 0.8 MHz from the degassed value ( $\Delta f$ ). Thus, inert gases were shown to interact with the nanotube-containing layer with a resulting change  
5 the dielectric constant of the system, changing the resonant frequency of the device. Another point that is evident from Figure 9 is that in the process of degassing the nanotubes, the Q-factor increased, i.e., the conductivity increased.

### **Example 2**

[0062] The response of the sensor of Example 1 was measured in oxygen.  
10 Figure 10 shows the response of the resonator to oxygen. The first two curves are the same as the corresponding curves in Figure 9. The third curve (circles) is the response of the sensor to oxygen. As can be seen, the resonant frequency of the resonator shifted from 3.8885 GHz in helium to 3.887 GHz in oxygen. This is very close to the resonant frequency value that was obtained in air. This strengthens  
15 the assumption that the electrical properties of pristine nanotubes are affected both by chemisorption and physisorption. This also reinforces the understanding that in as-prepared samples, the oxygen molecules have already interacted with the nanotubes and hence a similar shift was seen for air and oxygen gas exposures from the degassed value of resonant frequency.

20 [0063] Measurements were also carried out with argon and nitrogen and proportional shifts in the resonant frequency of the sensor circuit were observed, as can be seen in Figure 11. Thus, non-polar gases were shown to interact with pristine SWNTs and change their electrical properties. When as-prepared SWNT-containing sensors were exposed to these non-polar gases, no appreciable  
25 change in the resonant frequency of the sensor was detected.

### **Example 3**

[0064] Experiments were conducted on the degassed SWNT sensor circuits of Examples 1 and 2 to monitor their response to the presence of polar gases. Figure 8 shows the response of a SWNT-containing sensor to ammonia and  
30 carbon monoxide. The first curve (solid line) is the response of the sensor in air with a resonant frequency of 3.887 GHz. The second curve (squares) is the response of the sample after the nanotubes have been degassed. These are the same as the corresponding curves as shown in Figures 9 and 10. The third curve

(circles) is the response of the sensor circuit in carbon monoxide. The resonant frequency has shifted to 3.885875 GHz showing a downshift of 3.375 GHz. This shift is more than that of oxygen, which strengthens the belief that polar gases have a more pronounced effect on the dielectric properties of nanotubes. This can be due to the marked difference in the dielectric constants of polar and non-polar gases.

[0065] The fourth curve (crosses) shows the response of the SWNT sensor to ammonia. The resonant frequency shift in the case of ammonia exposure is slightly greater than that for carbon monoxide, which can be explained due to the relatively higher dielectric constant of ammonia with respect to CO, and also due to the larger amount of adsorption in the case of ammonia. The higher dielectric constant of ammonia is due to the higher dipole moment of the ammonia molecule and the larger amount of gas adsorption is believed to be due to the smaller molecular weight and higher adsorption energy of ammonia.

[0066] Another important fact that can be noticed in Figures 8-10 is the change in conductivity of the SWNT-containing sensor as it was exposed to different environmental conditions. As the as-prepared sample is degassed, the conductivity of the sensor increases, which is evident from the increase in the Q-factor of the sensor. Upon exposure of the sample to ammonia and carbon monoxide, the conductivity of the sensor decreases to a greater extent than when it was exposed to oxygen.

[0067] Figure 11 summarizes the final resonant frequency of the resonator under the different environmental conditions of Example 1-3. The plot shows the measurements of the resonant frequency in air, vacuum, and in different gases used. The solid line clearly demarcates the resonant frequency of the resonator in the presence of polar and non-polar gases. Among the polar gases, ammonia shows a greater shift in the resonant frequency due to its higher dielectric constant (higher dipole moment) compared to carbon monoxide. For non-polar gases, the shift follows the heats of adsorption more closely than the molecular weight as is evident due to the higher shift of nitrogen when compared to helium. This is believed to be due to the fact that the dielectric constants of these gases vary as the inverse of the square root of the molecular weight and directly as the exponential power of the heat of adsorption. Thus, there is a stronger dependence

of the resonant frequency shift on the heat of adsorption than molecular weight for non-polar gases.

#### **Example 4**

[0068] Electric-arc prepared SWNT bundles were physically coated on top  
5 of the conducting copper disk of a sensor as described in Examples 1-3.  
Electromagnetic energy was coupled into the resonator through a rectangular  
microstrip feed line. The SWNT layer was degassed by evacuating the testing  
chamber to  $10^{-5}$  torr and simultaneously heating the chamber to  $\sim 130^{\circ}\text{C}$  for about  
12 hours. The testing chamber was then cooled to room temperature while being  
10 maintained under high vacuum and the resonant frequency was recorded by a  
network analyzer before and after exposing the resonator to a known amount of  
gas or organic solvent vapor introduced into the chamber. The shift in resonant  
frequency was relatively large for vapors of acetone ( $\epsilon_r = 20$ ) compared to the  
corresponding shift for bromopropane ( $\epsilon_r = 8.1$ ) as seen by reference to Figures  
15 12A (illustrating the resonant frequency shift for acetone vapors) and 12B  
(illustrating the resonant frequency shift for bromopropane).

[0069] The shifts in the resonator frequency induced by the presence of  
 $\sim 1500$  ppm of different gaseous materials (gases or organic liquids) are  
summarized in Figure 14. These shifts ( $\Delta f$ ) scale proportionally as a function of  
20 the dielectric ( $\epsilon_r$ ) of the material, as indicated in the figures. All measurements  
were carried out under similar conditions using the same resonator for each of the  
materials shown in the Figure with a complete degassing cycle completed between  
exposures of the sensor to different materials. As can be seen, organic vapors  
with higher dielectric constants lead to a greater shift compared to the gases like  
25 nitrogen or oxygen, implying that the disclosed sensors can provide high selectivity  
for gas detection.

[0070] The change in resonant frequency for the resonator when exposed to  
a mixture of organic solvent vapors is also shown in Figure 14. Again, the change  
in resonant frequency ( $\Delta f$ ) for the mixture can be understood in terms of  $\epsilon_r$  for each  
30 component in the gas mixture. For mixtures containing more than two  
components, a detailed modeling for  $\Delta f$  in terms of  $\epsilon_r$  could be obtained by routine  
experimentation to compare with the experimentally observed values of  $\Delta f$ .

### **Example 5**

[0071] A resonant sensor including a layer containing degassed SWNT bundles such as that described for Example 4, above, was utilized to determine the response of the sensor to a gaseous flow including different concentrations of acetone vapors. The results are graphically illustrated in Figure 13. The first curve (squares) is the response of the sensor in vacuum with a resonant frequency at about 3.87 GHz. The second curve (triangles) shows the response of the sensor to a gaseous flow containing acetone at a concentration of 800 ppb. As can be seen, at this concentration, the resonant frequency shifts from the vacuum resonant frequency by 2 MHz. The third curve (circles) corresponds to the response of the sensor to a gaseous flow containing acetone at a concentration of 15 ppm, and demonstrates a resonant frequency shift from the vacuum resonant frequency of 8.6 MHz. The fourth curve (squares) corresponds to the response of the sensor to a gaseous flow containing acetone at a concentration of 1300 ppm, and demonstrates a resonant frequency shift from the vacuum resonant frequency of 44 MHz. In addition to the change in resonant frequency,  $\Delta f$ , another evident feature in Figure 13 is the decrease in the Q factor of the resonator upon exposure to different concentrations of acetone.

[0072] Similar to the unique resonant frequency of the sensors when in the presence of a known concentration of a given material, the change in resonant frequency of the sensors upon a change in concentration of a material will also be unique to that material. Thus, the disclosed sensors cannot only be utilized to identify a specific material in contact with the sensor, but, in certain embodiments, the disclosed sensors can also be utilized to determine the concentration of the material.

[0073] It will be appreciated that the foregoing examples, given for purposes of illustration, are not to be construed as limiting the scope of this invention. Although only a few exemplary embodiments of this invention have been described in detail above, those skilled in the art will readily appreciate that many modifications are possible in the exemplary embodiments without materially departing from the novel teachings and advantages of this invention. Accordingly, all such modifications are intended to be included within the scope of this invention which is defined in the following claims and all equivalents thereto. Further, it is

recognized that many embodiments may be conceived that do not achieve all of the advantages of some embodiments, yet the absence of a particular advantage shall not be construed to necessarily mean that such an embodiment is outside the scope of the present invention.